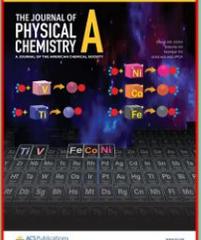
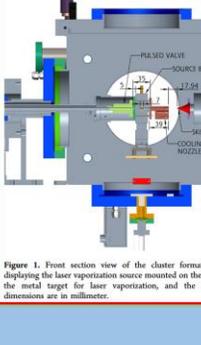


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<p>Research Focus and Scholarly Contributions of Professor Debashis Bandyopadhyay and Group (2021–2025): Renewable Energy, Hydrogen Storage, Sensors, Batteries, Catalysis, Materials, and Machine Learning</p>		
	<p>Ammonia Activation and Nitride Formation Pathways in Transition Metal Clusters: Insights from Mass Spectrometry and First-Principles DFT Vaibhav Chakraverty, Chaitanya Parasharath, Venu Vinayak Dandipati, Debashis Bandyopadhyay, and Soumen Bhattacharyya*</p>	<p>International Journal of Hydrogen Energy Hydrogen storage on MgO supported TiM_n (n=2-6) clusters: A first principle investigation</p>
<p>The role of oxygen defects in the electronic, optical and phonon dispersion of the LAGO perovskite: a density functional theory investigation</p>	<p>Investigating the stable structures of yttrium oxide clusters: Y_n clusters as promising candidates for O₂ dissociation</p>	<p>International Journal of Hydrogen Energy Unveiling reversible hydrogen storage mechanism on transition metal decorated 2D honey graphene: A density functional study</p>
<p>Hydrogen storage in Ti doped 4-6-8 biphenylene (TiL468): Insights from density functional theory</p>	<p>Insights into catalytic behavior of TiM_n (n=1-12) nanoclusters in hydrogen storage and dissociation process: A DFT investigation</p>	<p>International Journal of Hydrogen Energy Insights into the reversible hydrogen storage mechanism of transition metal-decorated Ir₁₂-graphene: A DFT study</p>
<p>Structure of small yttrium monoxide clusters, chemical bonding, and photoionization threshold: photoionization and density functional theory investigations</p>	<p>Unveiling Newly Modelled Janus Nb₂CO₂ MXene as a Promising Anode for Alkali Metal-Ion Batteries: A DFT Investigation</p>	<p>Materials Today Communications Insights into the electronic structure and stability of TiM_n (n = 1-12) clusters: Validation of electron counting rule</p>
<p>Ionization Energies and Ground-State Structures of Neutral La_n (n = 2-14) Clusters: A Combined Experimental and Theoretical Investigation</p>	<p>Physica E: Low-dimensional Systems and Nanostructures Study of electronic structure, stabilities and electron localization behavior of AgPb_n (n=1-14) nanoclusters: A first principal investigation</p>	<p>Machine Learning-Assisted Discovery of Perovskite Composites with Target Bandgaps</p>
 <p>Figure 1. Front section view of the cluster formation chamber displaying the laser vaporization source mounted on the PV, access to the metal target for laser vaporization, and the skimmer. All dimensions are in millimeter.</p>	<p>Materials Today Communications Insights into the electronic structure and stability of TiM_n (n = 1-12) clusters: Validation of electron counting rule</p>	<p>Collaborators</p> 

Focus: Single authorship, Invited review article with IF 16.8

Study of materials using Mössbauer spectroscopy, Debashis Bandyopadhyay, International materials reviews 51 (3) (2006) 171-208

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International Materials Reviews >
Volume 51, 2006 - Issue 3

Study of materials using Mössbauer spectroscopy

D. Bandyopadhyay*

A comprehensive review is presented of the recent contributions Mössbauer spectroscopy has made in materials science and engineering. After a brief introduction to the basic methodology, examples of the application of ^{57}Fe and ^{119}Sn Mössbauer spectroscopy in both transmission and back-scattering mode are presented and discussed. Recent technological and software developments of this technique are also included. Coverage is further extended to recent, pertinent developments in space research and also in biological science and technology where Mössbauer techniques are very widely used. Efforts have also been made to cover applications to archaeological samples where Mössbauer spectroscopy is an important analytical tool.

Keywords: Archaeology, Biological science, Magnetic materials, Metallic glass, Minerals, Mössbauer spectroscopy, Nanomaterials, Space research, Steels

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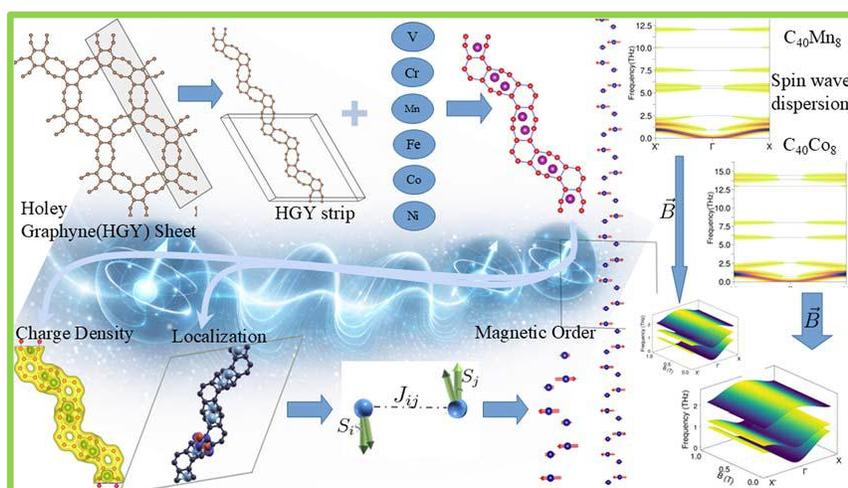
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Propagation of THz magnons in a one-dimensional transition metal decorated holey graphyne strip with tunable bandgaps



Prabin Pyakurel Sharma, ^a Chaithanya Purushottam Bhat ^a and Debashis Bandyopadhyay ^{*a}

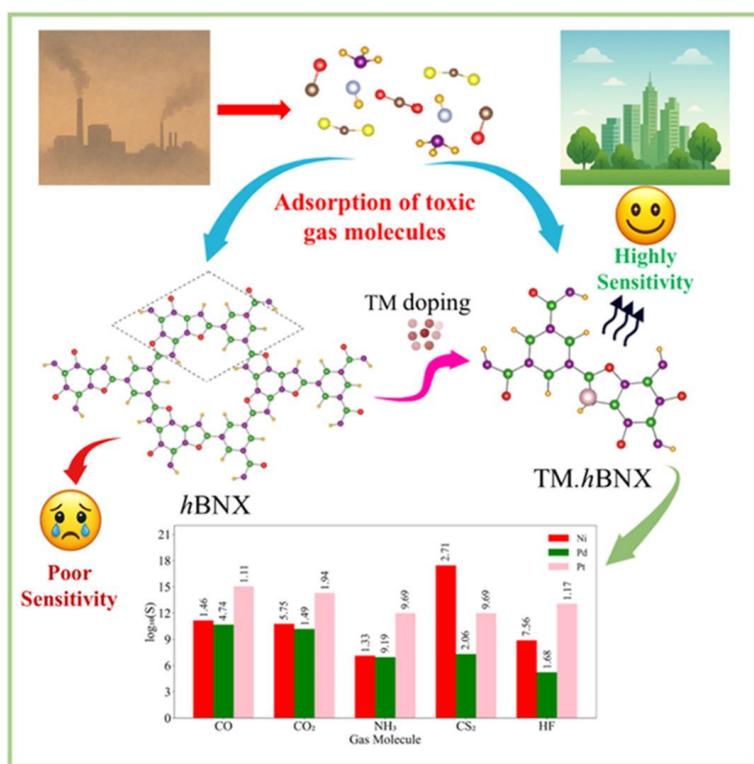
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Transition metal functionalized novel hBNX monolayer as toxic gas sensors: A density functional investigation

Chaithanya Purushottam Bhat ^a, Breeti Bandyopadhyay ^b, Brahmananda Chakraborty ^{c,d},
 Debashis Bandyopadhyay ^a  

Graphical Abstract



Chaithanya Purushottam Bhat, Pranav Suryawanshi, Aditya Guneja and Debashis Bandyopadhyay, "Unveiling the Adsorption and Electronic Interactions of Drugs on 2D Graphene: Insights from DFT and Machine Learning Approach", The article was first published on 23 Jan 2026 Mater. Adv., 2026, Accepted Manuscript, <https://doi.org/10.1039/D5MA01519G>

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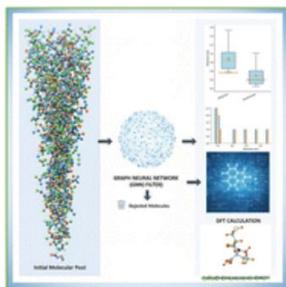


Cite this: *Mater. Adv.*, 2026,
7, 2803

Unveiling the adsorption and electronic interactions of drugs on 2D graphsene: insights from DFT and machine learning approaches

Chaithanya Purushottam Bhat, Pranav Suryawanshi, Aditya Guneja and Debashis Bandyopadhyay *

We developed a hybrid ML-DFT framework for rapid screening of drug-nanomaterial interactions (MAE 0.075 eV, DFT-validated). PDOS/Bader analyses reveal charge transfer/coupling mechanisms, enabling scalable ML-driven nanomaterial therapeutics design.



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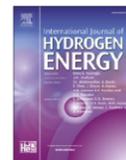
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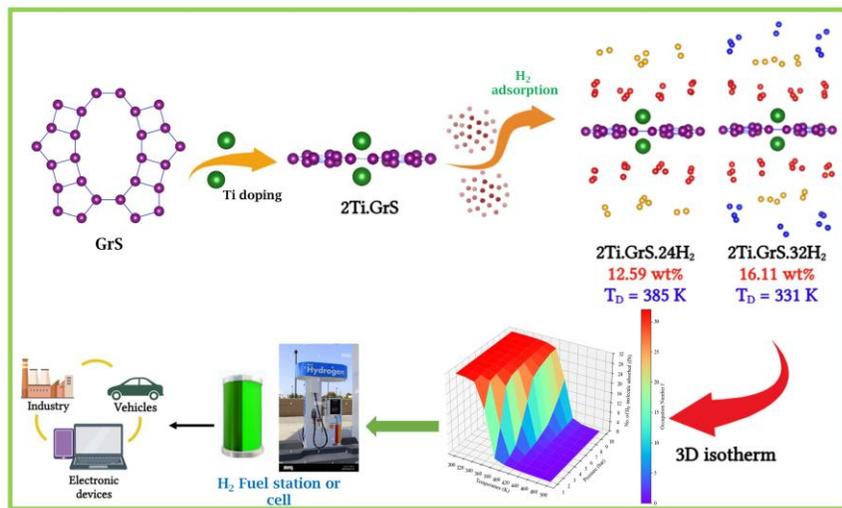
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Exploring reversible hydrogen storage on Ti- decorated novel 2D graphsene: A DFT investigation

Chaithanya Purushottam Bhat , Debashis Bandyopadhyay *

Department of Physics, Birla Institute of Technology and Science Pilani, Rajasthan, 333031, India

Graphical Abstract



66

Ammonia Activation and Nitride Formation Pathways in Transition Metal Clusters: Insights from Mass Spectrometry and First-Principles DFT

Vaibhav Chauhan, Chaithanya Purushottam Bhat, Varun Vinayak Deshpande, Debashis Bandyopadhyay, and Soumen Bhattacharyya*



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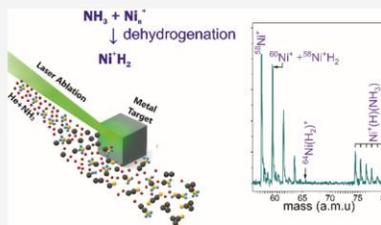
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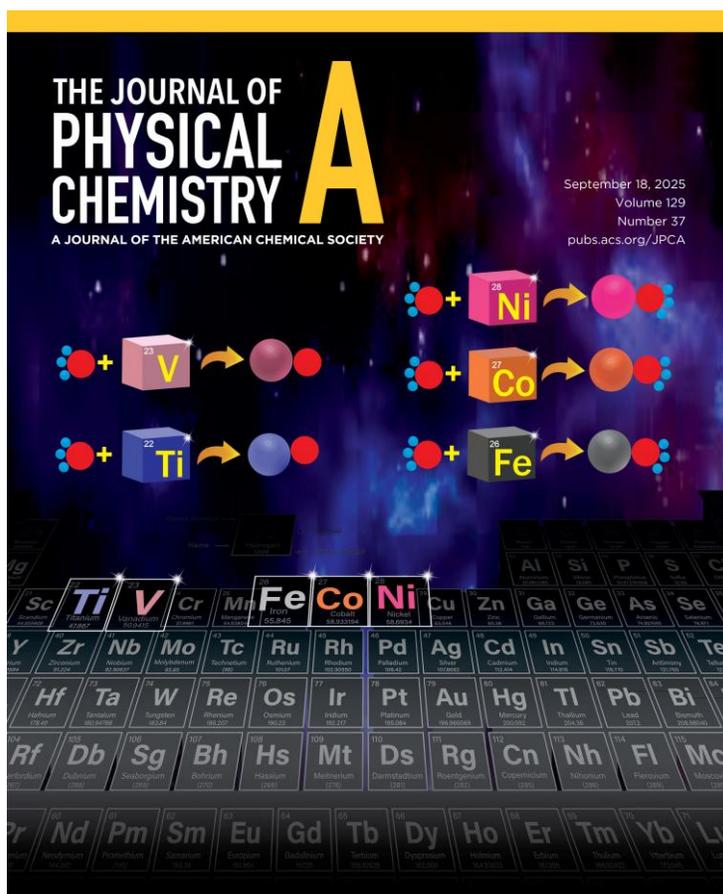
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ABSTRACT: The interaction of ammonia (NH_3) with laser-vaporized transition metal clusters (Ti, V, Fe, Co, and Ni) was systematically investigated using reflectron time-of-flight mass spectrometry and density functional theory. Metal-specific and size-dependent trends emerge: Ti clusters readily form $(\text{TiN})_n$ ($n = 1-7$), indicating strong nitride formation. Neutral V_n and Fe_n clusters predominantly yield mononitrides, with the NH_3 dehydrogenation efficiency varying with cluster size and charge. Co_n clusters show limited reactivity with mainly NH_3 adsorptions and partial dehydrogenation, while Ni_n clusters exhibit extensive NH_3 uptake, leading to stable nitride/imide species such as $\text{NiN}(\text{NH}_3)_4$ and $\text{Ni}(\text{NH})_2(\text{NH}_3)_4$, along with the formation of Ni^+H_2 via hydrogen release—likely resulting from the reaction of Ni_n^+ clusters with NH_3 . These findings provide insights into ammonia activation, N–H bond cleavage, and transition metal nitride formation mechanisms in small clusters.



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Investigating the structural, electronic, and magnetic properties of $\text{Fe}_2@\text{Ge}_n^\alpha$ ($\alpha = 0, +1, -1, n = 1-13$) nanoclusters: DFT insights

Ravi Trivedi^{a,b}, Vikash Mishra^c, Chaithanya Purushottam Bhat^d, Debashis Bandyopadhyay^{d,*}

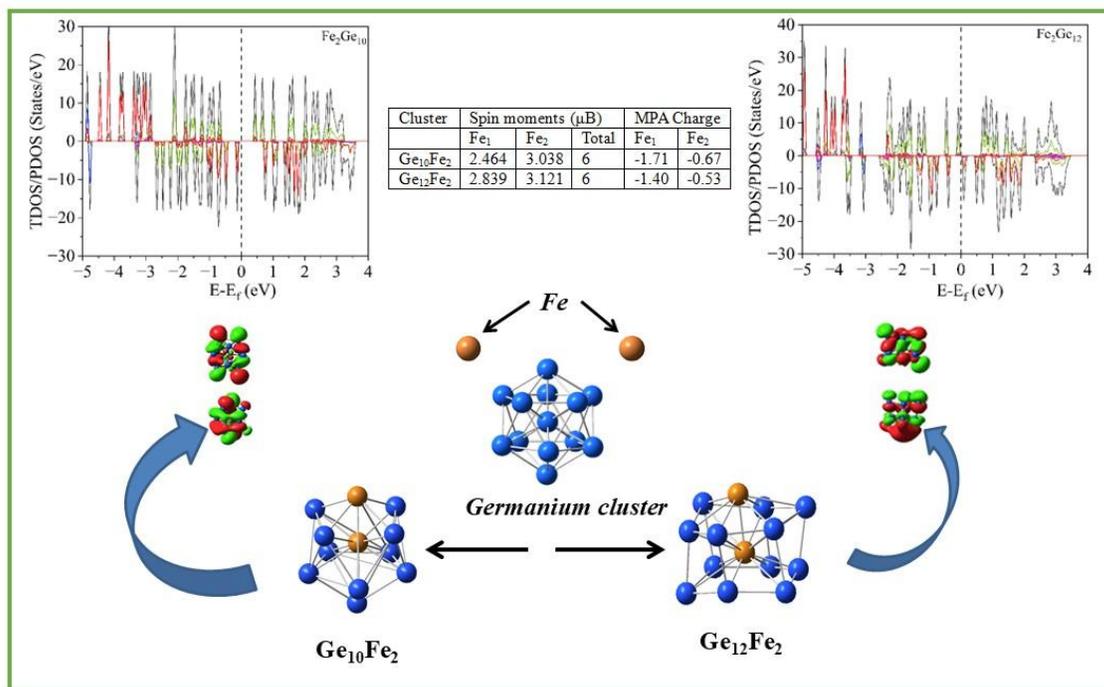
^a Department of Physics, Karpagam Academy of Higher Education, Coimbatore 641021 Tamil Nadu, India

^b Center for Computational Physics, Karpagam Academy of Higher Education, Coimbatore 641021 Tamil Nadu, India

^c Department of Physics, Manipal Institute of Technology, Manipal Academy of Higher Education, Manipal 576104 Karnataka, India

^d Department of Physics, Birla Institute of Technology and Science, Pilani, Rajasthan 333031, India

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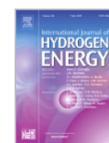
Unveiling reversible hydrogen storage mechanism on transition metal decorated 2D holey graphyne: A density functional study, Chaithanya Purushottam Bhat, Breeti Bandyopadhyay, Debashis Bandyopadhyay, *International Journal of Hydrogen Energy* 148 (2025) 150044, <https://doi.org/10.1016/j.ijhydene.2025.150044>

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International Journal of Hydrogen Energy

Volume 148, 16 July 2025, 150044



Unveiling reversible hydrogen storage mechanism on transition metal decorated 2D holey graphyne: A density functional study

Chaithanya Purushottam Bhat ^a, Breeti Bandyopadhyay ^b, Debashis Bandyopadhyay ^a

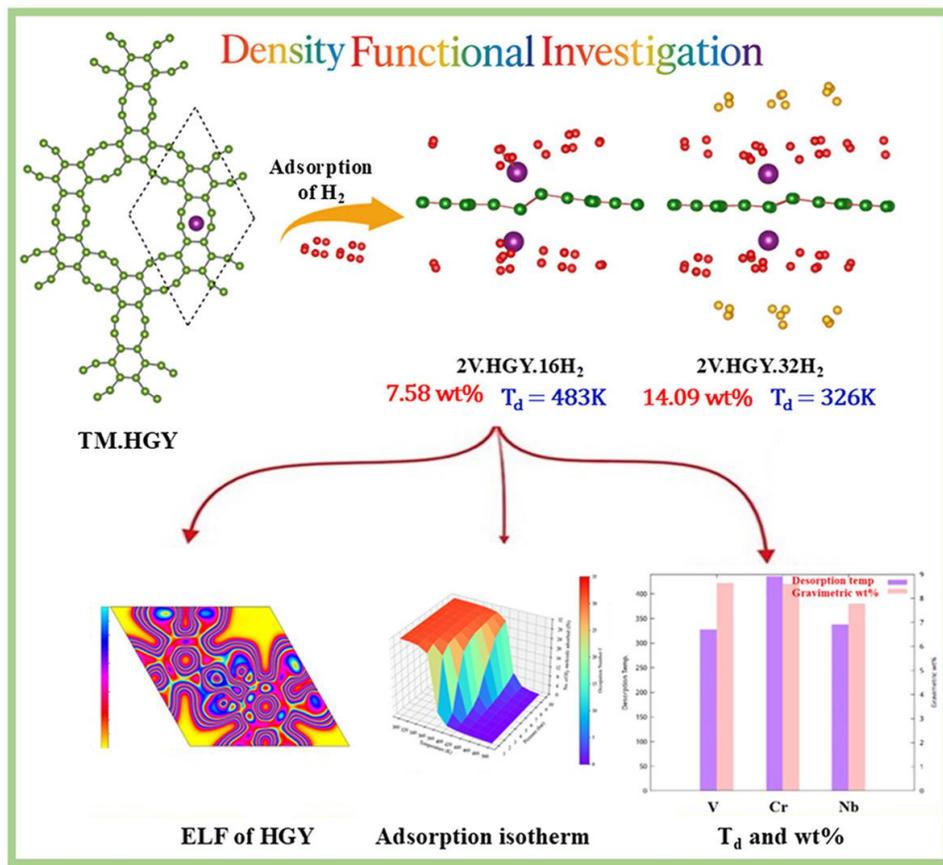
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<https://doi.org/10.1016/j.ijhydene.2025.05.072>

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International Journal of Hydrogen Energy

Volume 137, 12 June 2025, Pages 750-761



Insights into the reversible hydrogen storage mechanism of transition metal-decorated Irinda-graphene: A DFT study

Chaithanya Purushottam Bhat, Debashis Bandyopadhyay

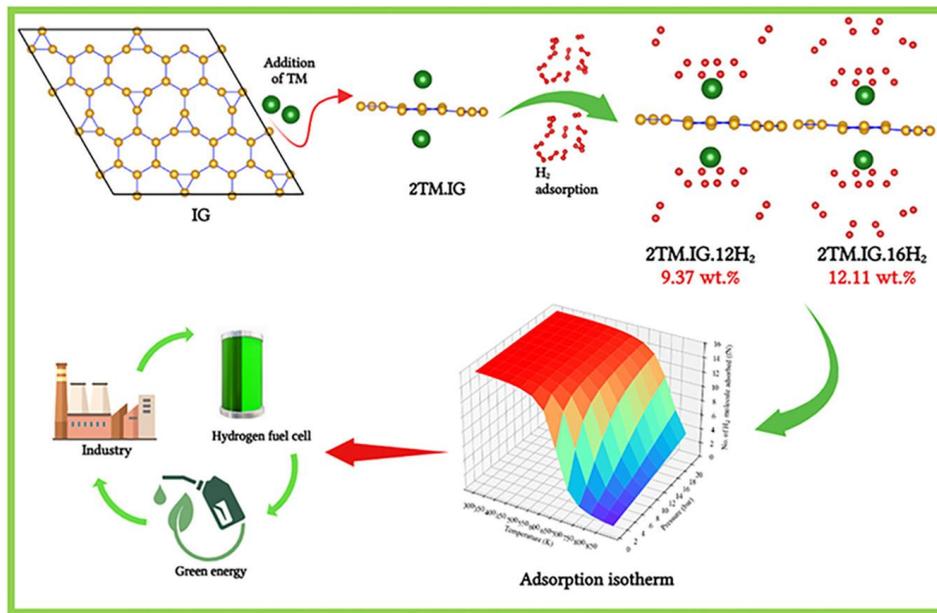
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A Novel 2D-hBNX Covalent Inorganic Framework Functionalized with Transition Metals for Enhanced Catechol Sensing: A Density Functional Investigation
 Chaithanya Purushottam Bhat, Debashis Bandyopadhyay, *Surfaces and Interfaces* 67 (2025) 106653,

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A Novel 2D-hBNX covalent inorganic framework functionalized with transition metals for enhanced catechol sensing: A density functional investigation

Chaithanya Purushottam Bhat, Debashis Bandyopadhyay

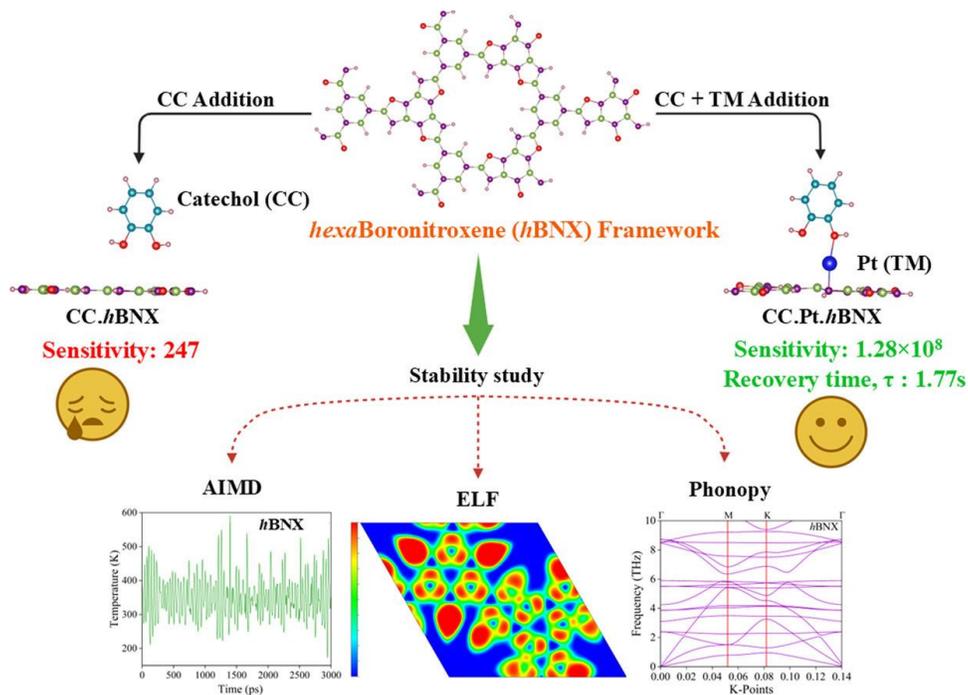
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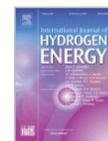
61 *Insights of Ti-doping on the hydrogen adsorption properties of the 2D-BeN₄ monolayer: A density functional investigation, CP Bhat, D Bandyopadhyay, International Journal of Hydrogen Energy 102 (2025) 1168-1179*

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Volume 102, 10 February 2025, Pages 1168-1179



Insights of Ti-doping on the hydrogen adsorption properties of the 2D-BeN₄ monolayer: A density functional investigation

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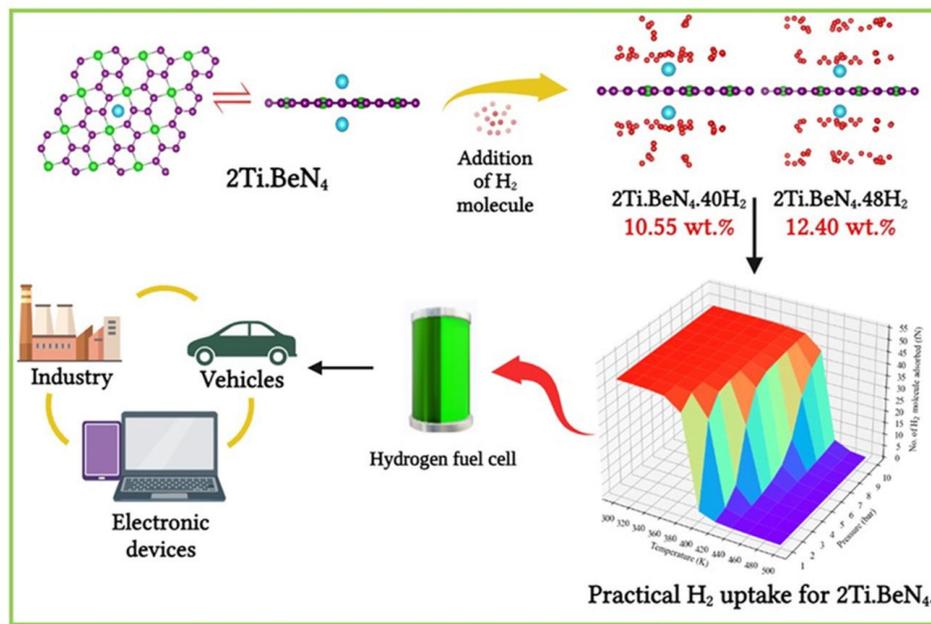
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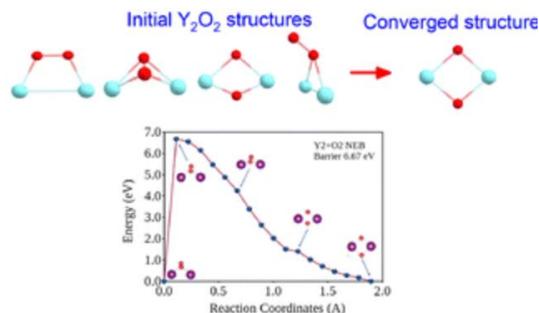


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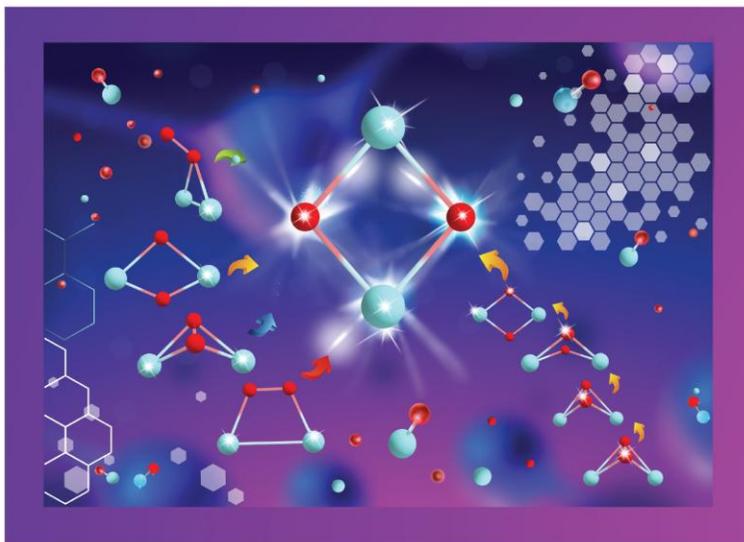
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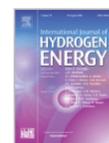
59 *Hydrogen storage in Ti doped 4-6-8 biphenylene (Ti. C468): Insights from density functional theory, Chaithanya Purushottam Bhat, Debashis Bandyopadhyay International Journal of Hydrogen Energy 79 (2025) 377-393, <https://doi.org/10.1016/j.ijhydene.2024.06.335>*

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International Journal of Hydrogen Energy

Volume 79, 19 August 2024, Pages 377-393



Hydrogen storage in Ti doped 4-6-8 biphenylene (Ti.C468): Insights from density functional theory

Chaithanya P. Bhat, Debashis Bandyopadhyay

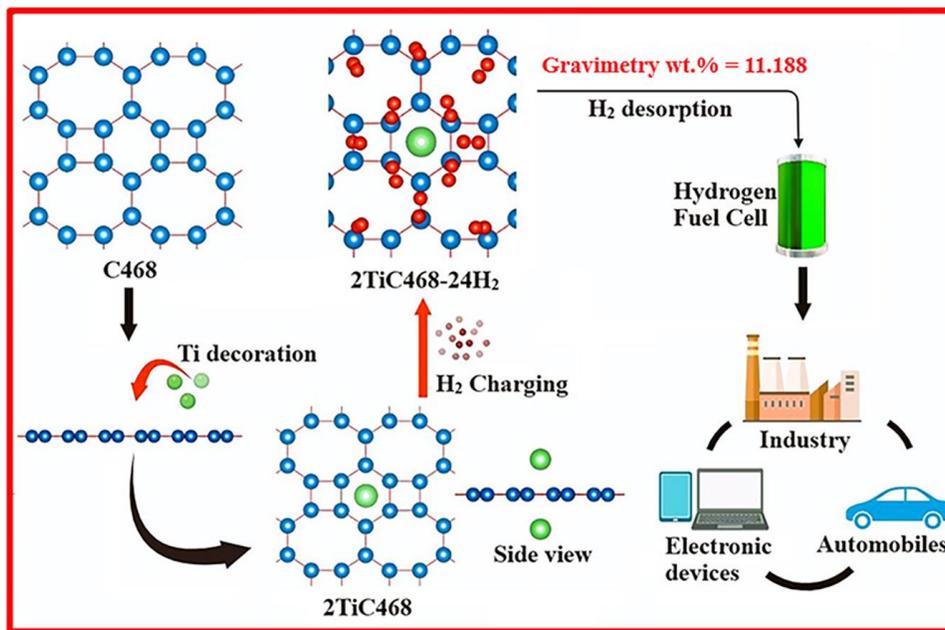
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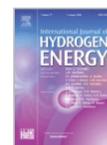
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Hydrogen storage on MgO supported TiMg_n (n= 2–6) clusters: A first principle investigation, S Chatterjee, D Bandyopadhyay, *International Journal of Hydrogen Energy* 77, (2024) 1467-1475,



International Journal of Hydrogen Energy

Volume 77, 5 August 2024, Pages 1467-1475



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Hydrogen storage on MgO supported TiMg_n (n=2–6) clusters: A first principle investigation

Soham Chatterjee, Debashis Bandyopadhyay

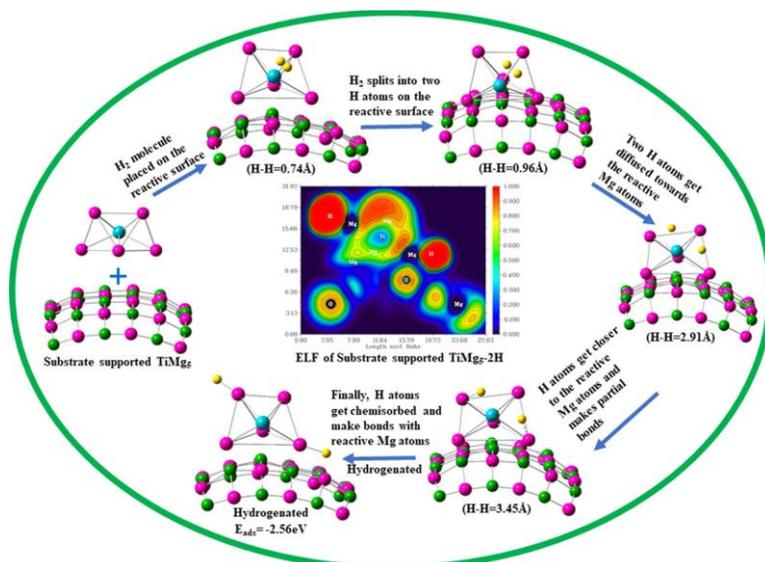
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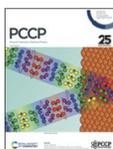
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Structure of small yttrium monoxide clusters, chemical bonding, and photoionization: threshold photoionization and density functional theory investigations, Varun Vinayak Deshpande, Vaibhav Chauhan, Debashis Bandyopadhyay, Anakuthil Anoop, Soumen Bhattacharyya, *Phys. Chem. Chem. Phys.*, 2024, 26, 20123-20133 <https://doi.org/10.1039/D4CP02351J>

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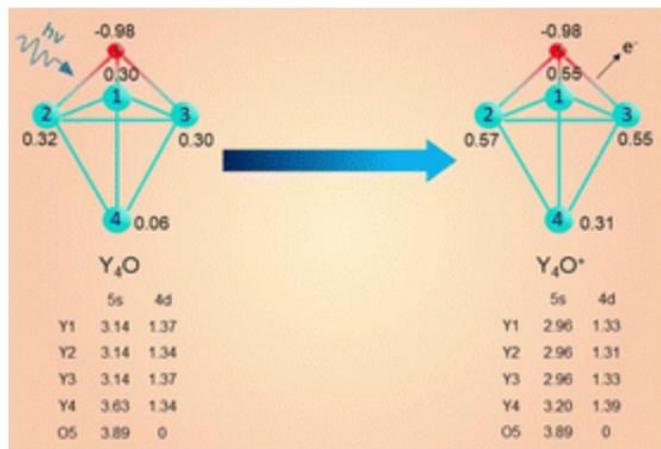
Structure of small yttrium monoxide clusters, chemical bonding, and photoionization: threshold photoionization and density functional theory investigations†



Varun Vinayak Deshpande,^{ab} Vaibhav Chauhan,^a Debashis Bandyopadhyay,^{id} ^c Anakuthil Anoop

^{id} ^d and Soumen Bhattacharyya ^{id} ^{*ab}

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56 *The role of oxygen defects in the electronic, optical and phonon dispersion of the LAGO perovskite: a density functional theory investigation*, Chaithanya Purushottam Bhat, Ashwin K Godbole, Debashis Bandyopadhyay, *Dalton Trans.*, 2023, 52, 16128-16139, DOI: 10.1039/d3dt02846a

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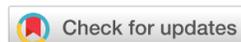
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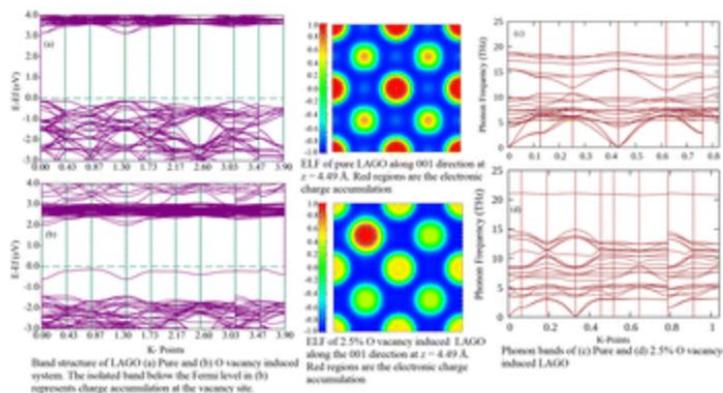


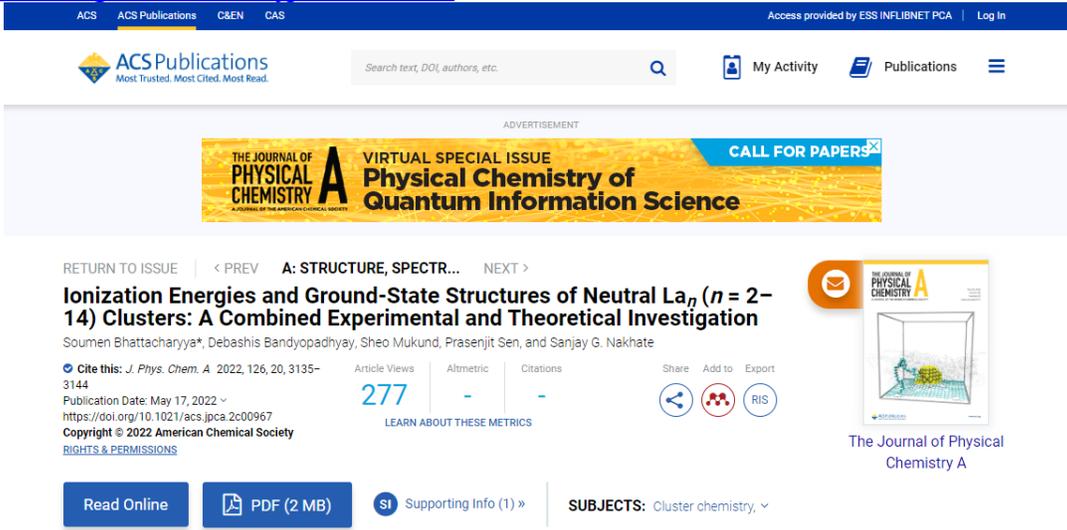
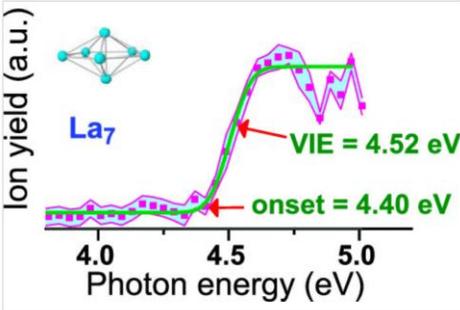
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The role of oxygen defects in the electronic, optical and phonon dispersion of the LAGO perovskite: a density functional theory investigation



Chaithanya P. Bhat,^a Ashwin K. Godbole^a and Debashis Bandyopadhyay ^{a*}



55	<p><i>First Principle Study on the Electronic and Optical Properties of KTaO₃ for Optoelectronic Properties, Trideeb Bhattacharya, Taha Yussuf Raja, Debashis Bandyopadhyay, 2023 IEEE Fifth International Conference on Advances in Electronics, Computers and Communications (ICAECC) (2023)Pages 01-04</i></p>	2023
54	<p><i>Ionization Energies and Ground-State Structures of Neutral La_n (n = 2–14) Clusters: A Combined Experimental and Theoretical Investigation</i> <i>S Bhattacharyya, D Bandyopadhyay, S Mukund, P Sen, SG Nakhate</i> <i>The Journal of Physical Chemistry A</i> 126 (20), (2022) 3135-3144 https://doi.org/10.1021/acs.jpca.2c00967</p>  <p>Abstract</p> <p>Neutral lanthanum clusters are studied by photoionization time-of-flight mass spectroscopy, laser threshold photoionization spectroscopy, and density functional theory (DFT). Mass abundance spectra (MS) registered at multiple photoionization wavelengths in the range of 195–230 nm by single photon ionization reveal the production of all sizes, La_n (n ≥ 50), in good abundance, nullifying previously predicted low abundances for certain sizes in the 3–14 size range. Also, the MS do not reveal the extraordinary stability of any specific size, as one would expect, from previous theoretical predictions of 7- and 13-atom clusters as magic. Ionization energies (IEs) are measured for La_n (n = 2–14) clusters. DFT has been used to determine the stable geometric isomers for 2- to 10-atom clusters and to calculate their IEs. The theoretical IEs of 2–7 atom clusters are in decent agreement with their experimental values; however, the theoretical IEs are somewhat lower by ~0.4 eV for n ≥ 8 than their experimental IEs.</p> 	2022 IF 2.7
53	<p><i>Insights into the electronic structure and stability of TiMgn (n= 1–12) clusters: Validation of electron counting rule, S Chatterjee, D Bandyopadhyay, Materials Today Communications</i> 32 (2022) 103860, https://doi.org/10.1016/j.mtcomm.2022.103860</p>	2022 IF 4.5

Insights into the electronic structure and stability of TiMg_n ($n=1-12$) clusters: Validation of electron counting rule

Soham Chatterjee, Debashis Bandyopadhyay

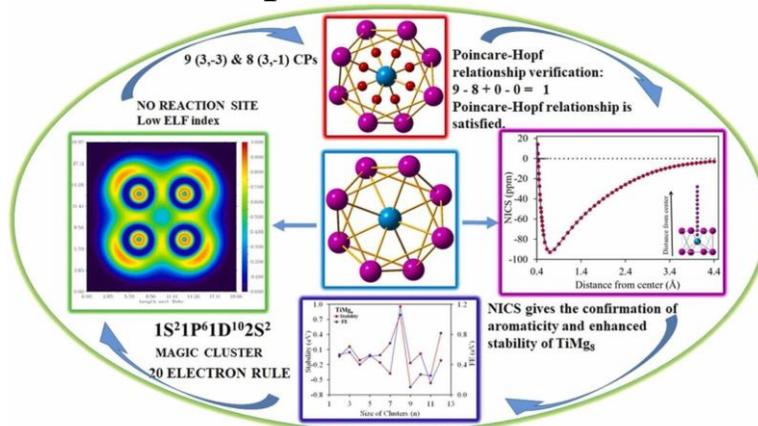
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52 *Insight into stabilities and magnetism of EuGen ($n = 1-20$) nanoclusters: an assessment of electronic aromaticity, Journal of Materials Science, 57 (2022) 19338–19355, <https://doi.org/10.1007/s10853-022-07834-0>*

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Insight into stabilities and magnetism of EuGe_n ($n = 1-20$) nanoclusters: an assessment of electronic aromaticity

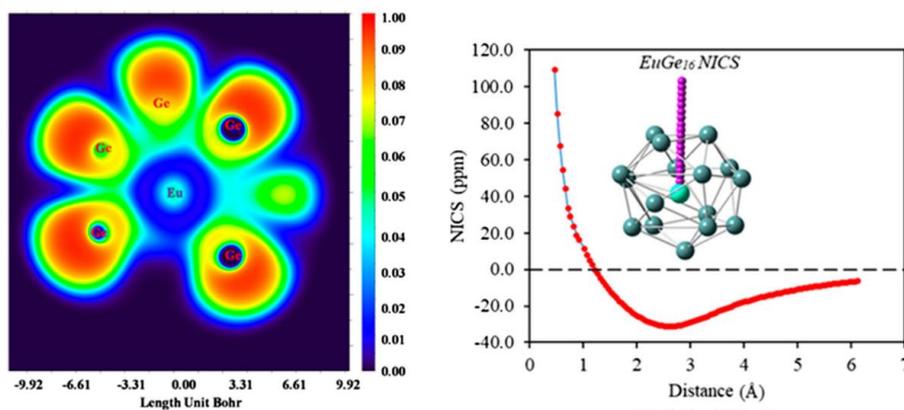
Ravi Trivedi¹, Antara Banerjee², and Debashis Bandyopadhyay^{3,*}

¹High Pressure and Synchrotron Radiation Physics Division, Bhabha Atomic Research Centre, Mumbai 400085, India

²Science Department, Vidya Niketan Birla Public School, Pilani, Rajasthan 333031, India

³Department of Physics, Birla Institute of Technology and Science, Pilani, Rajasthan 333031, India

GRAPHICAL ABSTRACT



ELF Mapping

ELF mapping and NICS show the presence of strong aromaticity in EuGe_{16} . Among the top 34 electrons, 18 σ and 16 π electrons (S.I.). The 18 σ electrons follow Hirsch's $2(n+1)^2$ σ -electron rule for $n=2$. The remaining 16 π electrons do not directly follow Hückel's $(4n+2)$ π -electron rule. Splitting it as $6\pi + 10\pi$ satisfies Hückel's rule for $n=1$ and 2, respectively. So, by applying the mixed π - σ electron counting rule, the enhanced stability of the EuGe_{16} cluster can explain.

51 Insights into catalytic behavior of TiMg_n ($n=1-12$) nanoclusters in hydrogen storage and dissociation process: A DFT investigation, **Debashis Bandyopadhyay**, Soham Chatterjee, Ravi Trivedi, and Kapil Dhaka, *Int. J. Hydrogen Energy*, 47(2022) 13418-13429, <https://doi.org/10.1016/j.ijhydene.2022.02.091>

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Insights into catalytic behavior of $TiMg_n$ ($n=1-12$) nanoclusters in hydrogen storage and dissociation process: A DFT investigation

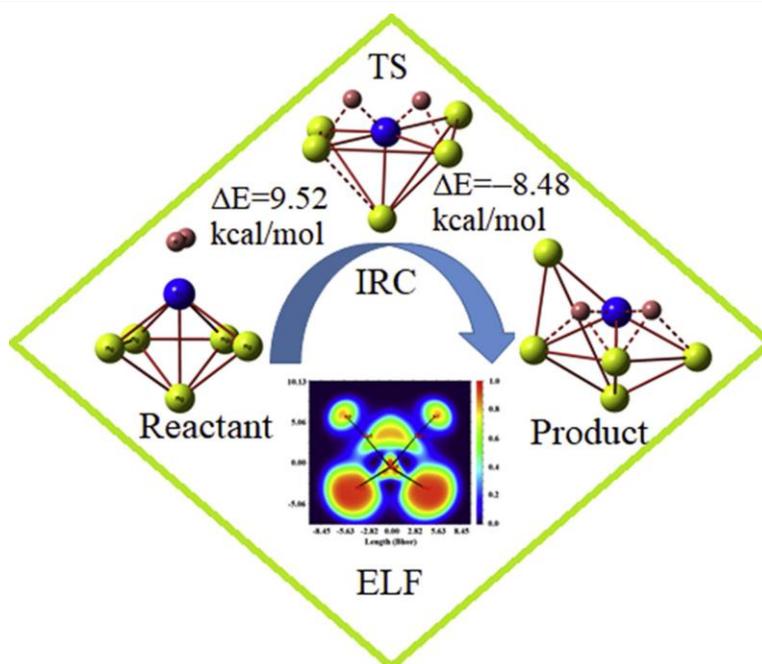
Debashis Bandyopadhyay ^a  , Soham Chatterjee ^a, Ravi Trivedi ^b, Kapil Dhaka ^c

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50 Study of electronic structure, stabilities and electron localization behavior of $AgPbn$ ($n=1-14$) nanoclusters: A first principal investigation. R Trivedi, A Banerjee, **Debashis Bandyopadhyay**, *Physica E: Low-dimensional Systems and Nanostructures* 131, 114725, <https://doi.org/10.1016/j.physe.2021.114725>

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Study of electronic structure, stabilities and electron localization behavior of AgPb_n ($n=1-14$) nanoclusters: A first principal investigation

Ravi Trivedi^a, Antara Banerjee^b, Debashis Bandyopadhyay^{c,*}

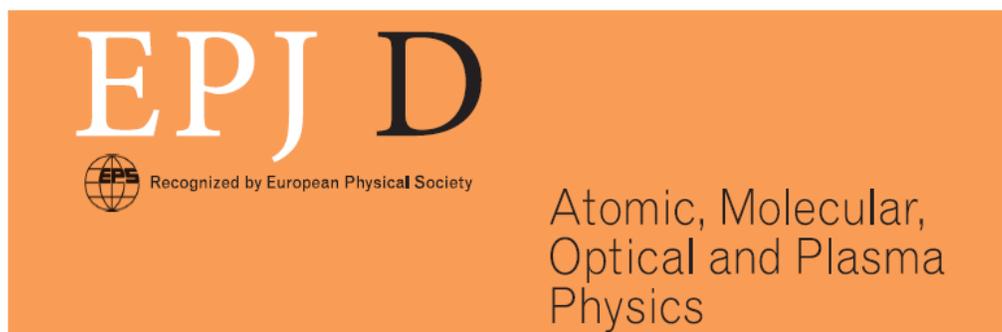
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^b Science Department, Vidya Niketan Birla Public School Pilani, Rajasthan, 333031, India

^c Department of Physics, Birla Institute of Technology and Science, Pilani, Rajasthan, 333031, India

- 49 *Ionization energies and structures of small lanthanum oxide clusters $(\text{La}_2\text{O}_3)_n\text{LaO}$ ($n = 1-3$), Soumen Bhattacharyya, Prasenjit Sen, Sheo Mukund, Suresh Yarlagadda, **Debashis Bandyopadhyay** and Sanjay G. Nakhate, *Eur. Phys. J. D* (2019) 73: 158 DOI: 10.1140/epjd/e2019-100185-5*

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Eur. Phys. J. D (2019) 73: 158

DOI: 10.1140/epjd/e2019-100185-5

Ionization energies and structures of small lanthanum oxide clusters $(\text{La}_2\text{O}_3)_n\text{LaO}$ ($n = 1-3$)

Soumen Bhattacharyya, Prasenjit Sen, Sheo Mukund, Suresh Yarlagadda, Debashis Bandyopadhyay, and Sanjay G. Nakhate

- 48 *Electronic structure and stability of anionic AuGe_n ($n=1-20$) clusters and assemblies: A density functional modelling, **Debashis Bandyopadhyay**, *Structural Chemistry*, (2019) 30: 955-963, DOI: 10.1007/s11224-018-1239-5, **Springer***

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Structural Chemistry (2019) 30:955-963
<https://doi.org/10.1007/s11224-018-1239-5>

ORIGINAL RESEARCH



Electronic structure and stability of anionic AuGe_n ($n = 1-20$) clusters and assemblies: a density functional modeling

Debashis Bandyopadhyay¹

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47	<p><i>Insights of the role of shell closing model and NICS in the stability of NbGe_n (n = 7-18) clusters: A first principle investigation, Ravi Trivedi and Debashis Bandyopadhyay, Journal of Materials Science, 2019 (54) 515-528, https://doi.org/10.1007/s10853-018-2858-3, IF. 4.22, Springer</i></p> <p><i>J Mater Sci</i></p> <p>Computation</p>  <p>Insights of the role of shell closing model and NICS in the stability of NbGe_n (n = 7–18) clusters: a first-principles investigation</p> <p>Ravi Kumar Trivedi^{1,2}  and Debashis Bandyopadhyay^{3,*} </p> <p>¹ Department of Theoretical Physics, Institute Ruder Boskovic, 10000 Zagreb, Croatia ² Present address: Department of Physics, Presidency University, Bengaluru, Karnataka 560064, India ³ Department of Physics, Birla Institute of Technology and Science, Pilani, Pilani, Rajasthan 333031, India</p>	2019 IF 3.5
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Study of adsorption and dissociation pathway of H₂ molecule on Mg_nRh (n = 1–10) clusters: A first principle investigation

Ravi Trivedi, Debashis Bandyopadhyay*

Department of Physics, Birla Institute of Technology and Science, Pilani, Pilani Campus, Rajasthan, 333031, India

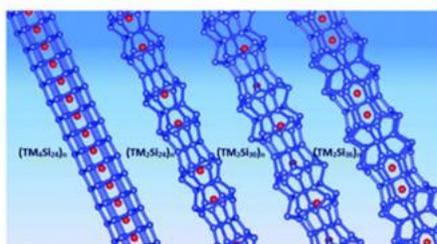
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**2016
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Magnetism, structures and stabilities of cluster assembled TM@Si nanotubes (TM = Cr, Mn and Fe): a density functional study

Kapil Dhaka and Debashis Bandyopadhyay

Four different types (Type 1 to Type 4) of empty and transition metal (Cr, Mn and Fe) doped silicon nanotubes have been studied. The calculated band structures and DOS assigned them as metallic, semiconductor, semi-metallic and half-metallic depending upon the combination of the type of nanotube and the transition metal doping.



The article was first published on 05 Jul 2016

Dalton Trans., 2016, **45**, 12432-12443<https://doi.org/10.1039/C6DT01252C>

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**2015
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4.6**

	<p style="text-align: right;"> RSC Advances</p> <p style="text-align: center; background-color: #4a7c9c; color: white; padding: 5px;">PAPER</p> <p> Cite this: <i>RSC Adv.</i>, 2015, 5, 83004</p> <p style="text-align: right;">Study of the electronic structure, stability and magnetic quenching of CrGe_n (n = 1–17) clusters: a density functional investigation†</p> <p style="text-align: right;">Kapil Dhaka and Debashis Bandyopadhyay*</p> <p>In the present report the evolution of the electronic structure, stability and magnetic quenching of CrGe_n nanoclusters has been carried out using density functional theory (DFT). From the nature of the variation of the different thermodynamic and chemical parameters, the CrGe₁₀ and CrGe₁₄ ground state clusters are identified as the most stable species. It is observed that the enhanced stability of CrGe₁₀ and CrGe₁₄ are due to the closed shell filled structure of the Cr-atomic orbitals and follow the 18-electron counting rule. It is found that the strong mixing of the Cr d-orbital with the s- and p-atomic orbitals of the Ge atoms in the cluster are mainly responsible for the stability and quenching of the Cr magnetic moment in the clusters. Calculated CPs also give additional information about the bonding and its effect on the stability of the clusters. Calculated IR and Raman spectra also support these results.</p> <p>Received 16th July 2015 Accepted 16th September 2015 DOI: 10.1039/c5ra13849c www.rsc.org/advances</p>	
42	<p style="text-align: center;">INTERNATIONAL JOURNAL OF HYDROGEN ENERGY 40 (2015) 12727–12735</p> <div style="display: flex; justify-content: space-between; align-items: center;">  <div style="text-align: center;"> <p>Available online at www.sciencedirect.com</p> <p style="font-size: 1.2em; font-weight: bold; color: green;">ScienceDirect</p> <p>journal homepage: www.elsevier.com/locate/he</p> </div>  </div> <p style="text-align: center;">Hydrogen storage in small size Mg_nCo clusters: A density functional study</p> <p style="text-align: center;">Ravi Trivedi, Debashis Bandyopadhyay*</p> <p style="text-align: center;"><small>Department of Physics, Birla Institute of Technology and Science, Pilani, Pilani Campus, Rajasthan, 333031, India</small></p> <p style="text-align: right;"> CrossMark</p>	2015 IF 8.3
41	<p style="text-align: right;"> RSC Advances</p> <p style="text-align: center; background-color: #4a7c9c; color: white; padding: 5px;">PAPER</p> <p style="text-align: right;">View Article Online <small>View Journal View Issue</small></p> <p> Cite this: <i>RSC Adv.</i>, 2014, 4, 64825</p> <p style="text-align: right;">Study of electronic properties, stabilities and magnetic quenching of molybdenum-doped germanium clusters: a density functional investigation†</p> <p style="text-align: right;">Ravi Trivedi, Kapil Dhaka and Debashis Bandyopadhyay*</p> <p>Evolution of electronic structures, properties and stabilities of neutral and cationic molybdenum encapsulated germanium clusters (Mo@Ge_n, n = 1 to 20) has been investigated using the linear combination of atomic orbital density functional theory method with effective core potential. From the variation of different thermodynamic and chemical parameters of the ground state clusters during the growth process, the stability and electronic structures of the clusters is explained. From the study of the distance-dependent nucleus-independent chemical shifts (NICS), we found that Mo@Ge₁₂ with hexagonal prism-like structure is the most stable isomer and possesses strong aromatic character. Density of states (DOS) plots of different clusters is then discussed to explain the role of d-orbitals of the Mo atom in hybridization. Quenching of the magnetic moment of the Mo atom with increase in the size of the cluster is also discussed. Finally, the validity of the 18-electron counting rule is applied to further explain the stability of the metallo-inorganic magic cluster Mo@Ge₁₂ and the possibility of Mo-based cluster-assembled materials is discussed.</p> <p>Received 5th October 2014 Accepted 3rd November 2014 DOI: 10.1039/c4ra11825a www.rsc.org/advances</p>	2014 IF 4.6

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	<p style="text-align: center;"><i>J. Phys. Chem. A</i> 2010, 114, 1835–1842 1835</p> <p style="text-align: center;">Density Functional Investigation of Structure and Stability of Ge_n and Ge_nNi (n = 1–20) Clusters: Validity of the Electron Counting Rule</p> <p style="text-align: center;">Debashis Bandyopadhyay¹ and Prasenjit Sen^{1*,2} <i>Physics Group, Birla Institute of Technology and Science, Pilani - 333031, Rajasthan, India, and Harish-Chandra Research Institute, Chhatnag Road, Jhansi, Allahabad-211019, U.P, India</i> <i>Received: June 14, 2009; Revised Manuscript Received: December 4, 2009</i></p> <p>Structure and electronic properties of neutral and cationic pure and Ni-doped Ge clusters containing 1–20 Ge atoms are calculated within the framework of linear combination of atomic orbitals density functional theory. It is found that in clusters containing more than 8 Ge atoms the Ni atom is absorbed endohedrally in the Ge cage. Relative stability of Ni-doped clusters at different sizes is studied by calculating their binding energy, embedding energy of a Ni atom in a Ge cluster, highest-occupied molecular orbital to lowest-unoccupied molecular orbital gap, and the second-order energy difference. Clusters having 20 valence electrons turn out to be relatively more stable in both the neutral and the cationic series. There is, in fact, a sharp drop in IP as the valence electron count increases from 20 to 21, in agreement with predictions of shell models. Relevance of these results to the designing of Ge-based superatoms is discussed.</p>	
32	<p><i>Effect of Transition Metal Doping on Hydrogenated Germanium Nanocages: A Density Functional Investigation, M Kumar, BJ Singh, S Kajjam, D. Bandyopadhyay</i> <i>Journal of Computational and Theoretical Nanoscience</i> 7 (1), 296-301</p> <p style="text-align: center;">Copyright © 2010 American Scientific Publishers All rights reserved Printed in the United States of America Journal of Computational and Theoretical Nanoscience Vol. 7, 296-301, 2010</p> <p style="text-align: center;">Effect of Transition Metal Doping on Hydrogenated Germanium Nanocages: A Density Functional Investigation</p> <p style="text-align: center;">Debashis Bandyopadhyay¹, Manish Kumar², Bandhan Jot Singh², and Shantanu Kajjam² ¹Physics Group²Electronics and Instrumentation Group, Birla Institute of Technology and Science, Pilani, Rajasthan 333031, India</p> <p>In this report we present an ab initio electronic-structure calculations of hydrogenated germanium cages Ge_nH_mTM (TM = Cu and Zn, n = 12 to 24) using density functional theory with polarized basis set (SDD) nanoclusters. In the first step of the calculation, geometrical optimizations of the nanoclusters have been done. In the next step only the ground state optimized geometries are used to calculate the binding energy (BE), HOMO-LUMO gap and embedding energy (EE) of the</p>	2010
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Study of pure and doped hydrogenated germanium cages: a density functional investigation

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Abstract

In this paper we present an *ab initio* electronic-structure calculation performed using density functional theory (DFT) with a polarized basis set (SDD) within the spin polarized generalized gradient approximation for pure and divalent transition metal doped hydrogenated germanium nanocluster cages $\text{Ge}_n\text{H}_n\text{M}$ ($M = \text{Zn}, \text{Cd}$ and Hg , $n = 6-28$). In the first step of the calculation, geometrical optimizations of the nanoclusters are done. In the next step only the ground state optimized geometries are used to calculate the binding energy (E_b), HOMO-LUMO gap (ΔE_g) and embedding energy of the clusters. To study the optical behaviour of the clusters, IR

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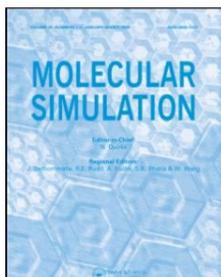
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The study of the electronic structures and properties of pure and transition metal-doped silicon nanoclusters: a density functional theory approach

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28 *The electronic structures and properties of transition metal-doped silicon nanoclusters: A density functional investigation, Debashis Bandyopadhyay, Manis Kumar, Chemical Physics* 353 (1) (2008) 170-176, <https://doi.org/10.1016/j.chemphys.2008.08.017>

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	<p style="text-align: center;">Chemical Physics 353 (2008) 170–176</p> <div style="display: flex; justify-content: space-between; align-items: center;">  <div style="text-align: center;"> <p>Contents lists available at ScienceDirect</p> <p>Chemical Physics</p> <p>journal homepage: www.elsevier.com/locate/chemphys</p> </div>  </div> <p style="text-align: center;">The electronic structures and properties of transition metal-doped silicon nanoclusters: A density functional investigation</p> <p style="text-align: center;">Debashis Bandyopadhyay*, Manish Kumar</p> <p style="text-align: center;"><small>Department of Physics, Birla Institute of Technology and Science, Pilani 333 031, Rajasthan, India</small></p> <hr/> <div style="display: flex; justify-content: space-between;"> <div style="width: 45%;"> <p>ARTICLE INFO</p> <p><small>Article history:</small> Received 4 April 2008 Accepted 20 August 2008 Available online 3 September 2008</p> <p><small>Keywords:</small> Ab initio DFT Nanoclusters Binding energy IR Raman IP EA</p> </div> <div style="width: 45%;"> <p>ABSTRACT</p> <p>We report an ab initio all electron molecular-orbital electronic-structure calculation by using density functional theory (DFT) and with polarized basis set (LanL2DZ) within the spin polarized generalized gradient approximation for metal-doped silicon clusters, Si_nM ($n = 14-20$ and $M = Ti, Zr, Hf$). As the first step of calculation, geometrical optimizations of the nanoclusters have been done. In the next step, these optimized geometries are used to calculate the binding energy and HOMO-LUMO gap (band gap) of the clusters. In order to check the dynamical stability of the clusters, IR and Raman spectra have been calculated. Further calculations have been done on cation and anion clusters to obtain ionization potential (IP), electron affinity (EA), chemical potential and chemical hardness of the optimized clusters.</p> <p style="text-align: right;"><small>© 2008 Elsevier B.V. All rights reserved.</small></p> </div> </div>	
27	<p><i>A density functional theory-based study of the electronic structures and properties of cage like metal doped silicon clusters, Debashis Bandyopadhyay, Journal of Applied Physics 104 (8) (2008) 4308, https://doi.org/10.1063/1.3000657</i></p> <p style="text-align: center;">JOURNAL OF APPLIED PHYSICS 104, 084308 (2008)</p> <p style="text-align: center;">A density functional theory–based study of the electronic structures and properties of cage like metal doped silicon clusters</p> <p style="text-align: center;">Debashis Bandyopadhyay^{a)} Department of Physics, Birla Institute of Technology and Science, Pilani Rajasthan 333031, India (Received 29 August 2007; accepted 1 September 2008; published online 23 October 2008)</p> <p><i>Ab initio</i> electronic-structure calculations were performed by using density functional theory with polarized basis set (LanL2DZ) within the spin polarized generalized gradient approximation for metal ($M=Ti,Zr,Hf$) doped Si_n clusters where n varies from 9 to 20. In the first step of the calculation, geometrical optimizations of the nanoclusters have been done. In the next step, these optimized geometries have been used to calculate the binding energy (BE) and HOMO-LUMO gap (ΔE_g) of the clusters. In order to check the stability of the clusters, the second order energy differences of the optimized geometries have been calculated. To study the optical behavior of the clusters, IR and Raman spectra calculation have been done. Further calculations on cation and anion clusters have been done to obtain their ionization potential (IP), electron affinity (EA), and chemical potential. © 2008 American Institute of Physics. [DOI: 10.1063/1.3000657]</p>	2008 IF 2.7
26	<p><i>Study of materials using Mössbauer spectroscopy, Debashis Bandyopadhyay, International materials reviews 51 (3) (2006) 171-208</i></p>	2006 IF 16.8

	<div style="display: flex; justify-content: space-between; align-items: center;">   </div> <h2 style="text-align: center;">Study of materials using Mössbauer spectroscopy</h2> <p>D. Bandyopadhyay*</p> <p>A comprehensive review is presented of the recent contributions Mössbauer spectroscopy has made in materials science and engineering. After a brief introduction to the basic methodology, examples of the application of ^{57}Fe and ^{119}Sn Mössbauer spectroscopy in both transmission and back-scattering mode are presented and discussed. Recent technological and software developments of this technique are also included. Coverage is further extended to recent, pertinent developments in space research and also in biological science and technology where Mössbauer techniques are very widely used. Efforts have also been made to cover applications to archaeological samples where Mössbauer spectroscopy is an important analytical tool.</p> <p><small>Keywords: Archaeology, Biological science, Magnetic materials, Metallic glass, Minerals, Mössbauer spectroscopy, Nanomaterials, Space research, Steels</small></p>	
25	<p><i>Study of kinetics of iron minerals in coal by ^{57}Fe Mössbauer and FT-IR spectroscopy during natural burning, Debashis Bandyopadhyay, Hyperfine interactions 163 (1-4) (2005) 167-176, https://doi.org/10.1179/174328006X79490</i></p> <p style="text-align: center;">Hyperfine Interactions (2005) 163: 167–176 DOI 10.1007/s10751-006-9227-3</p> <hr/> <h3 style="text-align: center;">Study of Kinetics of Iron Minerals in Coal by ^{57}Fe Mössbauer and FT-IR Spectroscopy During Natural Burning</h3> <p style="text-align: center;">Debashis Bandyopadhyay</p>	2005 IF 1.3
24	<p><i>The Ti-Si-C system (titanium-silicon-carbon), Debashis Bandyopadhyay, Journal of phase equilibria and diffusion 25 (5), 415-420</i></p> <p style="text-align: center;">JPEDAV (2004) 25:415-420 DOI: 10.1361/15477030420890 1547-7037/\$19.00 ©ASM International</p> <p style="text-align: right;">Basic and Applied Research: Section I</p> <hr/> <h3 style="text-align: center;">The Ti-Si-C System (Titanium-Silicon-Carbon)</h3> <p style="text-align: center;">Debashis Bandyopadhyay</p> <p style="text-align: center;">(Submitted September 12, 2003; in revised form June 15, 2004)</p> <p>In the present article different isothermals of Ti-Si-C system at temperatures ranging from 1250 to 2877 °C, previously reported by [1966Bru], [1989Tou], [1991Wak], and [1993Sei], were assessed and redrawn on the basis of the recently reported binary alloy phase diagram of Ti-Si, Ti-C, and Si-C.</p>	2004 IF 1.5
23	<p><i>Study of hyperfine field distributions and local magnetic order of $\text{Fe}_{80-x}\text{Ni}_x\text{Cr}_{20}$ alloys by ^{57}Fe Mossbauer spectroscopy, Debashis Bandyopadhyay, ICAMMP-2002: International Conference on Advances in Materials Processing ...</i></p>	2002
22	<p><i>The C-Ti-Zr System (Carbon–Zirconium-Titanium), D. Bandyopadhyay, RC Sharma, N Chakraborti J. Phase Equilibria and Diffusion 22 (1), 61</i></p>	2001 IF 1.5

	<p>Phase Diagram Evaluations: Section II</p> <hr/> <h2 style="text-align: center;">The C-Ti-Zr System (Carbon-Titanium-Zirconium)</h2> <p style="text-align: center;"><i>D. Bandyopadhyay, R.C. Sharma, and N. Chakraborti, Indian Institute of Technology</i></p> <div style="display: flex; justify-content: space-around;"> <div style="width: 45%;"> <p>Ti-C System</p> <p>The assessed phase diagram of the Ti-C system in Fig. 1 is taken from [1998Oka], [1995Alb], [1996Jon], and [1996Sei] have reported other assessments of this system; all show two terminal solids α-Ti and β-Ti and a refractory monocarbide TiC with other phases being liquid and graphite (C). Two eutectic reactions and one peritectoid reaction take place in this system at 1646 °C, 2776 °C, and 920 °C, respectively. There seems to be a tendency of carbon ordering at composi-</p> </div> <div style="width: 45%;"> <p>Ti-Zr System</p> <p>[1969Rud], [1982Auf], [1982Mur], and [1987Mur] have studied the phase diagrams of the Ti-Zr system. The assessed phase diagram of the Ti-Zr system shown in Fig. 2 is taken from [1994Har], [1995Oka] provides a comparison between the Ti-Zr phase diagram given by [1987Mur] and [1994Har]. [1987Mur] calculated the phase diagram on the basis of the data given by [1982Auf]. In Ti-Zr system, the high-temperature bcc β-modifications and low-temperature hcp α-modifi-</p> </div> </div>	
21	<p><i>Calculation of the Debye temperature and study of the lattice dynamics of $Fe_{80-x}Ni_xCr_{20}$ by ^{57}Fe Mössbauer spectroscopy, D. Bandyopadhyay, RM Singru, AK Majumdar, Zeitschrift für Metallkunde 92 (4), 367-369</i></p>	2001
20	<p>Phase Diagram Evaluations: Section II</p> <hr/> <h2 style="text-align: center;">The C-Hf-Ti System (Carbon-Hafnium-Titanium)</h2> <p style="text-align: center;"><i>D. Bandyopadhyay, R.C. Sharma, and N. Chakraborti, Indian Institute of Technology</i></p> <div style="display: flex; justify-content: space-around;"> <div style="width: 45%;"> <p>Ti-C System</p> <p>The assessed phase diagram of the Ti-C system shown in Fig. 1 is taken from [1998Oka], [1995Alb], [1996Jon], and [1996Sei] reported other assessments of this system, which consists of two terminal solids α-Ti and β-Ti and a refractory monocarbide TiC. The other phases present are liquid and graphite (C). Two eutectic and one peritectoid reaction appear in this system at 1646, 2776, and 920 °C, respectively. There seems to be a tendency of carbon ordering at compositions below stoichiometry; as a result, the Ti₂C phase does not appear in the binary phase diagram. The crystal structure</p> </div> <div style="width: 45%;"> <p>C-Hf System</p> <p>[1990Oka] conducted a detailed assessment of the C-Hf system, where only one intermediate phase exists. The assessed phase diagram shown in Fig. 3 is taken from [1990Oka]. Three terminal solids, C (graphite), α-Hf, and β-Hf, are present in this system. [Massalski] has reported the temperature of β-Hf \leftrightarrow α-Hf allotropic transformation as 1743 °C. There are several conflicting reports in the literature ([1954Cot], [1961Por], [1962Kat], and [1965Rud]) regarding the eutectic reaction temperature between C and Hf and its</p> </div> </div>	2000 IF 1.5
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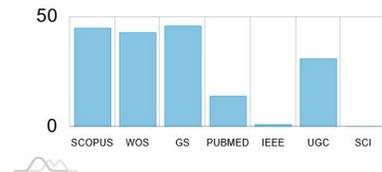
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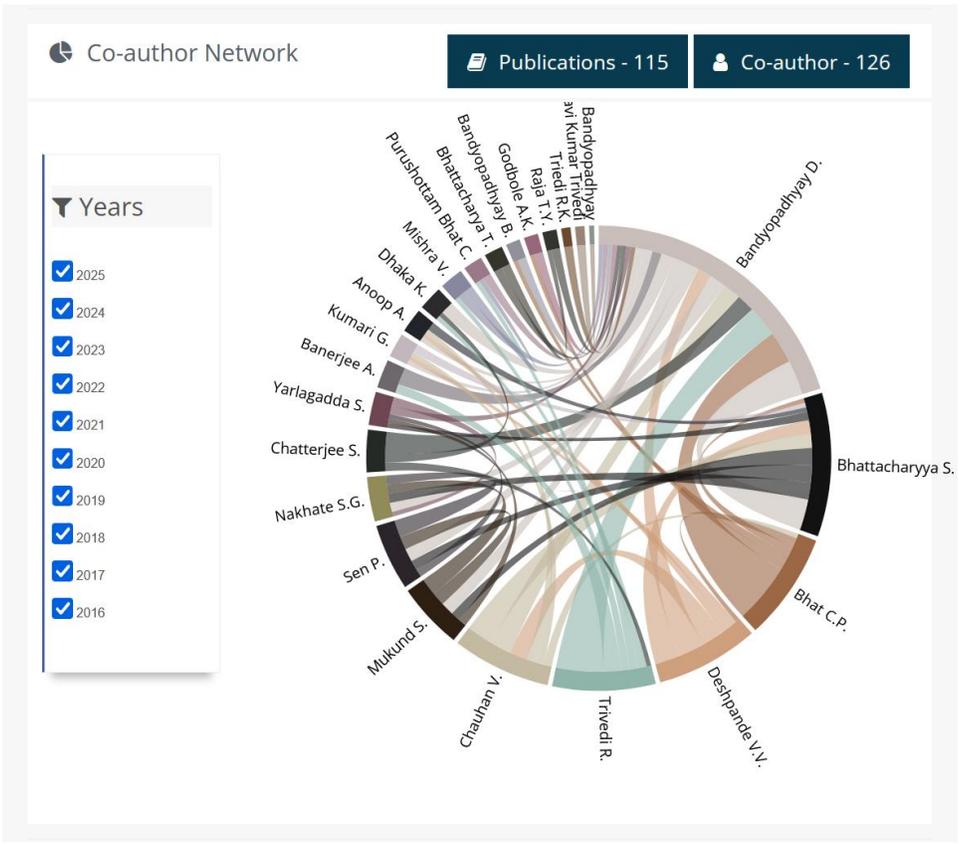
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